THE EFFECT OF SILVER AND BISMUTH DOPED Mg₂Si ON CRYSTAL STRUCTURE AND THERMOELECTRIC PROPERTIES

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ABSTRACT

Magnesium silicide have been high performance in temperature ranges 500 - 800 K which are medium-temperature and appropriate for applications with waste heat. The Ag and Bi doped Mg2Si were synthesized by hot press method. The crystal structure was analyzed by X-ray diffraction. Thermoelectric properties were measured in temperature rang 323 - 473 K. The XRD pattern of all samples show single crystal and crystallite size slightly increase for doped sample. The doped samples show thermal conductivity lower than un-dope and Ag-doped show maximum decrease about 44.44% at 473 K. The electrical resistivity of doped samples was decreased, and decrease 33% for Bi doped at 473 K. The doped materials slightly decreased Seebeck coefficient about 4% and 16% for Ag and Bi doped, respectively. The highest ZT was found in Ag doped about 0.47×10^{-2} at 473 K.

KEYWORDS: Alternative Energy; Thermoelectric; Magnesium silicide; Silver; Bismuth

INTRODUCTION

Thermoelectric (TE) materials has been investigated in recent years for renewable heat energy into electricity. The performance of TE materials were confirmed by dimensionless figure of merit $ZT=S^2T/\sigma\kappa$, where S is Seebeck coefficient, T is temperature, σ is electrical resistivity and κ is thermal conductivity). Good TE materials needed to high Seebeck coefficient, electrical resistivity low and thermal conductivity. Magnesium silicide have been high performance in temperature ranges 500-800 K which are medium-temperature and appropriate for applications with waste heat. The highest about 1.4 for the ZT was reported $Mg_2(Si_{0.4}Sn_{0.6})Sb_{0.018}$ sample preparing by twice mechanical alloying and spark plasma sintering process [1], $Mg_2Si_{0.53}Sn_{0.4}Ge_{0.05}Bi_{0.02}$ preparing by solid state synthesis and sintering via hot pressing [2], $Mg_{2.16}(Si_{0.4}Sn_{0.6})_{0.97}Bi_{0.03}$ preparing by a two-step solid state reaction followed by spark plasma sintering. [3] and $Mg_2Si_{0.55}Sn_{0.4}Ge_{0.05}:Bi=1:0.02$ preparing by powder methods [4]. This research has objective for improve ZT of Mg_2Si by doped with Ag and Bi.

MATERIALS AND METHODS

The Ag and Bi doped Mg_2Si were prepared by solid state reaction and hot press method. Mg (99%, Aldrich), Si (99.9%, Aldrich), Ag (99.9%, Aldrich) and Bi (99%, Aldrich) were used as starting raw materials. The raw materials were mixed by ball milling for 12 h and calcined at

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973 K for 1 h in Ar atmosphere. The powder placed in the graphite mold (ϕ 20 mm) for hot press method at 1053 K under pressure 33 MPa for 1 h in Ar atmosphere.

The crystal structure was characterized by X-ray diffraction (XRD; Shimadzu 6100, Japan) method using CuK α radiation at 40 kV, 30 mA with a scanning speed of 5°/min at 2 θ steps of 0.02°. The morphology was observed by scanning electron microscope (SEM; JEOL JSM–5401, Germany). Seebeck coefficient, electrical resistivity and thermal conductivity were measured by steady state method [5] at temperature ranges of 323 – 473 K.

RESULTS AND DISCUSSION

The X-ray diffraction analysis of Mg₂Si, Mg_{1.98}Bi_{0.02}Si and Mg_{1.98}Ag_{0.02}Si after hot press are compare with PDF#2 00-035-0773 as shown in Fig. 1. The crystal structure of all samples are show cubic Fm3m space group as a major phase, and the minor secondary phase MgO and Si are detected in all compositions due to the oxidation of Mg during the solid state reaction [6]. The SEM images magnification \times 3500 show distribution, size and shape of particle. The particle size show 2–6 µm and evenly distribution as show in Fig. 2.



Fig. 1 XRD pattern of Mg₂Si (blue line) Mg_{1.98}Bi_{0.02}Si (red line) and Mg_{1.98}Ag_{0.02}Si (green line).



Fig. 2 SEM images of (a) Mg_2Si (b) $Mg_{1.98}Bi_{0.02}Si$ and (c) $Mg_{1.98}Ag_{0.02}Si$.

The Seebeck coefficient increases with temperature increasing but decreased when doping with Ag and Bi. The Mg₂Si shows highest value about -150μ V/K at 473 K as shown in Fig. 3.

Temperature dependence of electrical resistivity for Mg_2Si , $Mg_{1.98}Bi_{0.02}Si$ and $Mg_{1.98}Ag_{0.02}Si$ shows in Fig. 4. The Ag and Bi doped effect to decease electrical resistivity due to increase in carrier concentration and Bi doped shows lowest value about 60 m Ω cm at 473 K.



Fig. 3 Seebeck coefficient of Mg_2Si , $Mg_{1.98}Bi_{0.02}Si$ and $Mg_{1.98}Ag_{0.02}Si$ depend on temperature.



Fig. 4 Electrical resistivity of Mg_2Si , $Mg_{1.98}Bi_{0.02}Si$ and $Mg_{1.98}Ag_{0.02}Si$ depend on temperature.



Fig. 5 Thermal conductivity of Mg_2Si , $Mg_{1.98}Bi_{0.02}Si$ and $Mg_{1.98}Ag_{0.02}Si$ depend on temperature.

Temperature dependence of thermal conductivity for Mg_2Si , $Mg_{1.98}Bi_{0.02}Si$ and $Mg_{1.98}Ag_{0.02}Si$ shows in Fig. 5. The doped samples can be decease thermal conductivity and

Ag doped shows lowest thermal conductivity about 3 W/m K at 473 K as shown in Fig. 3 (c).

The Fig. 6 shows dimensionless figure of merit of Mg_2Si , $Mg_{1.98}Bi_{0.02}Si$ and $Mg_{1.98}Ag_{0.02}Si$. Doped samples show ZT value higher than undope and the Ag doped shows maximum ZT value about 0.47×10^{-2} at 473 K.



Fig. 6 Dimension less figure of merit of Mg_2Si , $Mg_{1.98}Bi_{0.02}Si$ and $Mg_{1.98}Ag_{0.02}Si$ depend on temperature.

CONCLUSION

The doped samples show thermal conductivity lower than un-dope and Ag doped show maximum decrease about 44.44% at 473 K. The electrical resistivity of doped samples was decreased, and decrease 33% for Bi doped at 473 K. The doped materials slightly decreased Seebeck coefficient about 4% and 16% for Ag and Bi doped, respectively. The highest ZT was found in Ag doped about 0.47×10^{-2} at 473 K.

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